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AN IMPROVED PREPARATION OF 3-(5-BENZOFURANYL)-L-ALANINE Giorgio Ortar^a

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(175); 111 (9%); 99 (100%); 83 (19%); 71 (27%); 55 (31%); 43 (35%). HRMS: Calcd for $C_{16}H_{30}O_2$, *m/z*, 254.22458. Found: 254.22857

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Submitted by	Giorgio	Ortar
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3-(5-Benzofuranyl)-L-alanine (3a) has recently emerged as a highly promising non-canonical amino acid in the development of strategies for the site-specific *in vivo* incorporation of photoreactive amino acids.¹ The synthesis of 3a in five steps and ~10% overall yield has involved the nonselective preparation of the D,L-form followed by enzymatic resolution. In the last fifteen years, a large variety of 3-aryl and 3-heteroarylalanines have been prepared as valuable tools in the design of selective peptide ligands exploiting transition metal-catalyzed reactions.² In particular, Crisp has used the Sonogashira reaction³ to attach a series of fluorescent and enzymatic labels containing a terminal alkyne to suitably protected iodophenylalanines.⁴ Annulation of Sonogashira adducts of *o*-halophenols and 1-alkynes, either as starting materials or as reaction intermediates, represents a very useful procedure for the synthesis of benzofurans.⁵ The use of a similar chemistry to obtain the title compound from commercially available 3-iodo-L-tyrosine is described herein.



The reaction of fully protected 3-iodo-L-tyrosine $1a^6$ with (trimethylsilyl)acetylene in the presence of Pd(OAc)₂/PPh₃ as the catalyst, CuI as cocatalyst, and Et₃N as the base in DMF proceeded smoothly at 45° to give the protected 3-ethynyl-L-tyrosine 1b in 90% yield. Less satisfactory results were obtained at 60° using piperidine as the base (56%)^{5d} or at 90° in Et₃N (74%)^{5c}. No benzofuran 2a resulting from cyclization of 1b could be observed in these syntheses.⁷ Desilylation of 1b with KF in MeOH and *in situ* cyclization of 1c by the catalytic action of CuI at room temperature provided protected 3-(5-benzofuranyl)-L-alanine 2b in 83% yield. When Pd(OAc)₂ was used as catalyst in the cyclisation step, 2b was isolated in only 19% yield. The free amino acid was obtained nearly quantitatively by saponification of 2b with a mixture NaOH/MeOH/H₂O at room temperature followed by the removal of the Boc protecting group using a solution of dry HCl in ethyl acetate. Physical data for the hydrochloride salt 3b (mp, optical rotation, ¹H and ¹³C NMR) were in good agreement with the values reported in literature.¹

EXPERIMENTAL SECTION

Melting points were determined on a Kofler hot-stage apparatus and are uncorrected. Optical rotations were measured at 25° with a Schmidt-Haensch Polartronic D polarimeter (1 dm-cell). IR spectra were recorded on a Perkin-Elmer 983 spectrophotometer. ¹H and ¹³C NMR spectra were obtained on a Varian Mercury 300 spectrometer.

N-Boc-3-[(trimethylsilyl)ethynyl]-L-tyrosine Methyl Ester (1b).- A mixture of $1a^6$ (421 mg, 1 mmol), (trimethylsilyl)acetylene (0.21 mL, 1.5 mmol), Et₃N (0.56 mL, 4 mmol), Pd(OAc)₂ (7 mg, 0.03 mmol), PPh₃ (16 mg, 0.06 mmol), and CuI (11 mg, 0.06 mmol) in DMF (4 mL) was stirred at 45° for 3 h under nitrogen. The reaction mixture was then cooled, diluted with water, and extracted with Et₃O. The organic solution was washed twice with water, dried (Na₂SO₄), and evaporated. The residue

(459 mg) was purified by chromatography on silica gel (14 g) using $CH_2Cl_2/hexane = 7/3$ as eluent to give 352 mg (90%) of **1b**: oil; $[\alpha]_D$ +39° (c 1.0, CHCl_3); IR (CHCl_3): 3517, 3437, 2953, 2145, 1742, 1710, 1488, 1366, 1164 cm⁻¹; ¹H NMR (CDCl_3): δ 0.27 (9H, s, SiMe_3), 1.43 (9H, s, *t*-Bu), 2.92 (1H, dd, J = 14.0, 6.3 Hz, C<u>H</u>H), 3.03 (1H, dd, J = 14.0, 5.6 Hz, CH<u>H</u>), 3.72 (3H, s, CO_2Me), 4.51 (1H, m, α -CH), 4.99 (1H, d, J = 8.1 Hz, NH), 5.82 (1H, br s, OH), 6.86 (1H, d, J = 8.1 Hz, ArH), 6.99 (1H, dd, J = 8.1, 2.1 Hz, ArH), 7.11 (1H, br s, ArH); ¹³C NMR: δ 0.05, 28.35, 37.41, 52.30, 54.55, 80.06, 98.90, 102.45, 109.70, 114.81, 127.86, 131.59, 132.39, 155.09, 156.24, 172.31.

Anal. Calcd for C₂₀H₂₀NO₅Si: C, 61.35; H, 7.47; N, 3.58. Found: C, 61.26; H, 7.54; N, 3.63

N-Boc-3-(5-benzofuranyl)-L-alanine Methyl Ester (2b).- To a stirred solution of 1b (391 mg, 1 mmol) in MeOH (4 mL) was added KF (116 mg, 2 mmol) and the resulting mixture was stirred at 20-25°. The desilylation reaction followed by TLC (CH₂Cl₂/AcOEt = 9/1, R_r of 1b and 1c 0.43 and 0.32, respectively) was complete in 6 h. CuI (19 mg, 0.1 mmol) was added and the stirring was continued at 20-25° for additional 36 h. The reaction mixture was then diluted with water and extracted with Et₂O. The organic solution was washed with water, dried (Na₂SO₄), and evaporated. The residue (308 mg) was purified by chromatography on silica gel (10 g) using CH₂Cl₂/hexane = 7/3 as eluent to give 265 mg (83%) of 2b, R_r 0.72 (CH₂Cl₂/AcOEt = 9/1): oil; $[\alpha]_D$ +45° (c 1.0, CHCl₃); IR (CHCl₃): 3437, 2977, 1740, 1707, 1602, 1500, 1367, 1170 cm⁻¹; ¹H NMR (CDCl₃): δ 1.41 (9H, s, *t*-Bu), 3.17 (2H, m, β-CH₂), 3.71 (3H, s, CO₂Me), 4.61 (1H, m, α-CH), 5.03 (1H, d, J = 7.6 Hz, NH), 6.71 (1H, d, J = 2.2 Hz, ArH); ¹³C NMR: δ 28.36, 38.32, 52.25, 54.88, 80.01, 106.50, 111.42, 121.78, 125.60, 127.82, 130.53, 145.44, 154.27, 155.21, 172.52.

Anal. Calcd for C₁₇H₂₁NO₅: C, 63.94; H, 6.63; N, 4.39. Found: C, 63.83; H, 6.70; N, 4.45

3-(5-Benzofuranyl)-L-alanine Hydrochloride (3b).- A solution of 2b (319 mg, 1 mmol) in MeOH (3 mL) and 2N NaOH (1 mL) was stirred at 20-25° for 2 h, then concentrated under reduced pressure. acidified with a slight excess of 2N HCl, and extracted with AcOEt. The organic phase was washed twice with water, dried (Na_2SO_4) , and evaporated. The residue (316 mg) was purified by flash chromatography on silica gel using CH₂Cl₂/MeOH = 95/5 as eluent to give 298 mg (98%) of 3c, R_{f} 0.41 $(CH_2Cl_2/MeOH = 9/1)$: oil; $[\alpha]_{D_1} + 43^\circ$ (c 1.0, CHCl_3); IR (CHCl_3): 3434, 2938, 1706, 1500, 1369, 1164 cm⁻¹; ¹H NMR (CDCl₃): δ 1.41 (9H, s, t-Bu), 3.17 (1H, dd, J = 13.5, 5.7 Hz, C<u>H</u>H), 3.28 (1H, dd, J = 13.5, 5.7 Hz, CH<u>H</u>), 4.64 (1H, m, α -CH), 5.00 (1H, d, J = 7.5 Hz, NH), 6.71 (1H, d, J = 2.1 Hz, ArH), 6.97 (1H, br s, CO₂H), 7.11 (1H, d, J = 8.4 Hz, ArH), 7.40 (1H, br s, ArH), 7.42 (1H, d, J = 8.4 Hz, ArH), 7.60 (1H, d, J = 2.1 Hz, ArH); ¹³C NMR: δ 28.22, 37.62, 54.59, 80.18, 106.27, 111.21, 121.63, 125.40, 127.54, 130.02, 145.09, 153.96, 154.90, 175.95. To a solution of 3c (298 mg, 0.98 mmol)) in AcOEt (2 mL) was added a saturated solution of anhydrous HCl in AcOEt (4 mL). The solution was stirred at 20-25° for 3 h and evaporated to afford 3b as a white solid (234 mg, 99%), mp 225-226°; $[\alpha]_{D}$ -14° (c 1.0, H₂O) [*lit.*¹ mp 234-236°; $[\alpha]_{D}$ -12.9° (c 1.0, H₂O)]; IR (KBr): 3420, 2901, 1734, 1483, 1212 cm⁻¹; ¹H NMR (D₂O): δ 3.29 (1H, dd, J = 14.7, 7.5 Hz, C<u>H</u>H), 3.45 (1H, dd, J = 14.7, 5.5 Hz, CH<u>H</u>), 4.32 (1H, dd, J = 7.5, 5.5 Hz, α -CH), 6.90 (1H, d, J = 2.1 Hz, ArH), 7.26 (1H, d,

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J = 8.4 Hz, ArH), 7.57 (1H, d, J = 8.4 Hz, ArH), 7.59 (1H, br s, ArH), 7.79 (1H, d, J = 2.1 Hz, ArH); ¹³C NMR: δ 35.43, 54.58, 106.35, 111.66, 121.98, 125.29, 127.87, 128.41, 146.23, 153.98, 171.72.

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- o-Alkynylphenols are known to cyclise under basic conditions (refs. 5a,b), but some authors have reported that the reaction does not proceed to an appreciable extent without transition metals catalysis (ref. 5c). Indeed, we found that **1b** could be converted into **2a** in 44% yield by treatment with Et₃N (2 equivalents) and a catalytic amount of Pd(OAc)₂ (5 mol%) in MeOH at 45° for 4.5 h.
 2a: oil; [α]_D +38° (c 1.0, CHCl₃); IR (CHCl₃): 3437, 2953, 2927, 1740, 1707, 1498, 1368, 1252, 1164 cm⁻¹; ¹H NMR (CDCl₃): δ 0.34 (9H, s, SiMe₃), 1.41 (9H, s, *t*-Bu), 3.17 (2H, m, β-CH₂), 3.71 (3H, s, CO₂Me), 4.59 (1H, m, α-CH), 4.97 (1H, d, J = 5.6 Hz, NH), 6.90 (1H, d, J = 0.6 Hz, ArH), 7.02 (1H, dd, J = 8.5, 1.6 Hz, ArH), 7.31 (1H, br s, ArH), 7.41 (1H, d, J = 8.5 Hz, ArH); ¹³C NMR: δ 1.83, 30.12, 39.95, 53.99, 56.62, 81.72, 113.04, 117.68, 123.28, 127.44, 130.23, 131.71, 157.02, 159.23, 166.00, 174.28.
 Anal. Calcd for C₂₀H₂₉NO₅Si: C, 61.35; H, 7.47; N, 3.58. Found: C, 61.25; H, 7.55; N, 3.65 2-(Trimethylsilyl)benzofurans have been previously obtained by flash vacuum pyrolysis of *o*-[(trimethylsilyl)ethynyl]phenols: T. J. Barton and B. L. Groh, *J. Org. Chem.*, **50**, 158 (1985) and by metallation of benzofurans followed by quenching with Me₃SiCl [M. Gill, *Tetrahedron*, **40**,

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